

Supplementary Information

Enhanced green upconversion luminescence in tetrahedral LiYF₄:Yb/Er nanoparticles by manganese(II)-doping: The key role of host lattice

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Table S1. Crystallographic parameters of cubic NaYF₄ and tetragonal LiYF₄ crystals.

Phase	Atom	Site	x/a	y/b	z/c
Tetragonal-LiYF₄ (I4_{1/a})¹	Li	4a	0	1/4	1/8
	Y	4b	1/2	3/4	1/8
	F	16f	0.2815(1)	0.0854(1)	0.0437(1)
Cubic-NaYF₄ (Fm-3m)²	Na	4a	0	0	0
	Y	4a	0	0	0
	F	8c	1/4	1/4	1/4

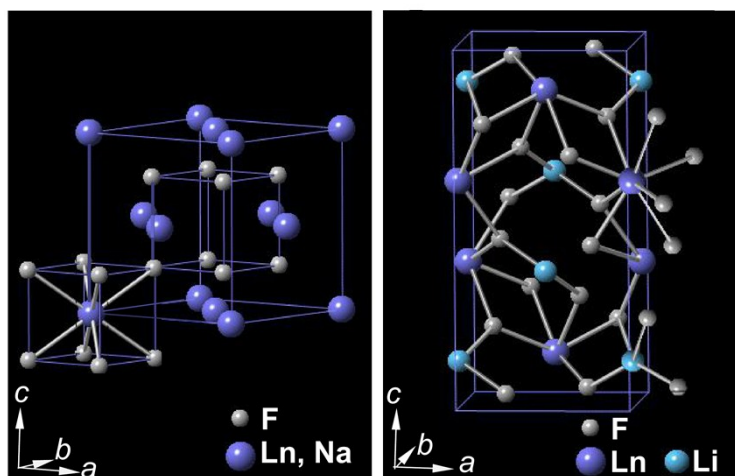


Figure S1. Schematic representation of cubic-phase NaYF₄ (left) and tetragonal-phase LiYF₄ (right) structure.

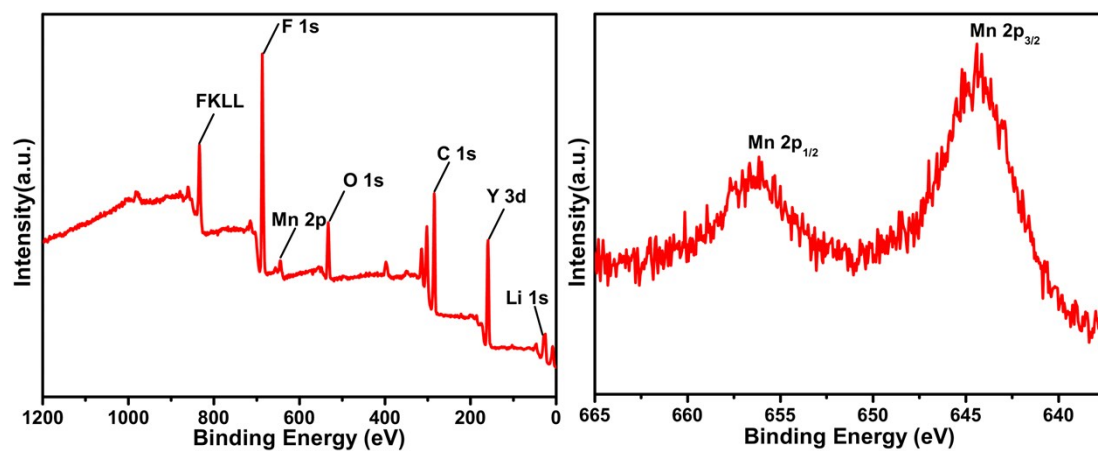


Table S2. Elemental analyses of Mn²⁺ in Mn²⁺ doped LiYF₄:Yb/Er nanoparticles by ICP-AES measurement.

Nominal concentration (mol%)	ICP-AES result (mol%)
5	0.46
8	0.48
10	0.72
12	0.83
14	1.21

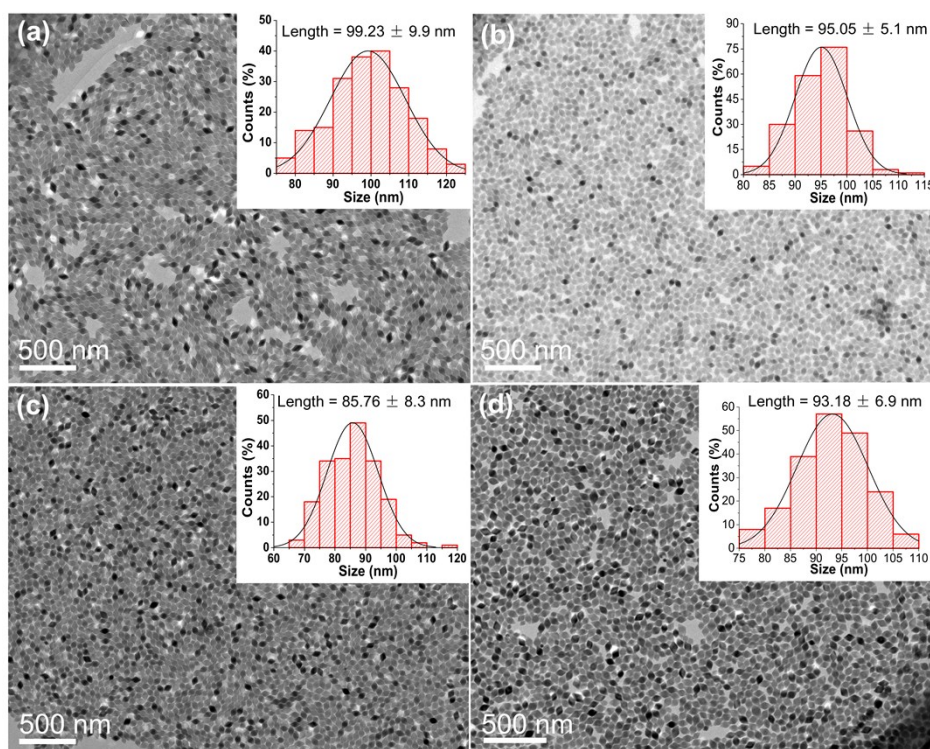


Figure S3. Low-resolution TEM images and the corresponding size distributions of (a) $\text{LiYF}_4:\text{Yb/Er/Mn}$ (20/2/0 mol%), (b) $\text{LiYF}_4:\text{Yb/Er/Mn}$ (20/2/5 mol%), (c) $\text{LiYF}_4:\text{Yb/Er/Mn}$ (20/2/10 mol%) and (d) $\text{LiYF}_4:\text{Yb/Er/Mn}$ (20/2/14 mol%) nanocrystals. The size distributions of the nanocrystals were calculated by counting over 200 particles recorded in the TEM images.

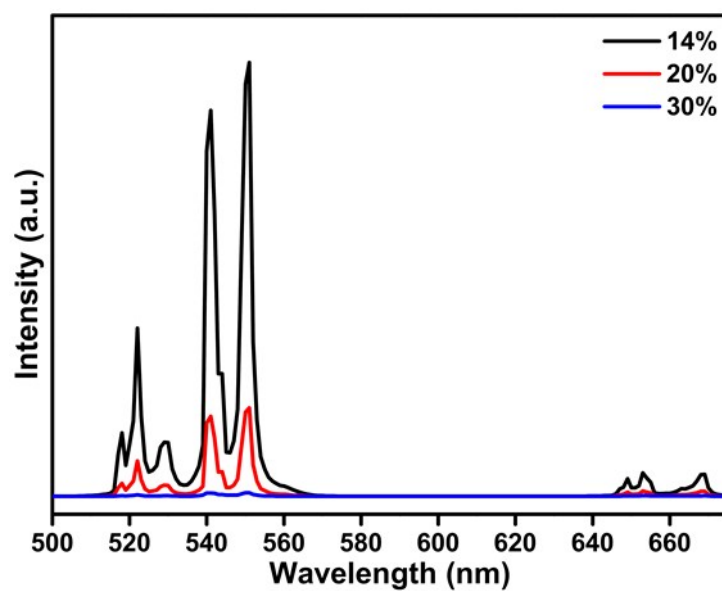


Figure S4. Room-temperature UC emission spectra of cyclohexane solutions containing LiYF₄:Yb/Er/Mn (20/2/14-30 mol%) nanoparticles with different Mn²⁺ doping levels, under 980 nm laser excitation (CW, 2 W·cm⁻²).

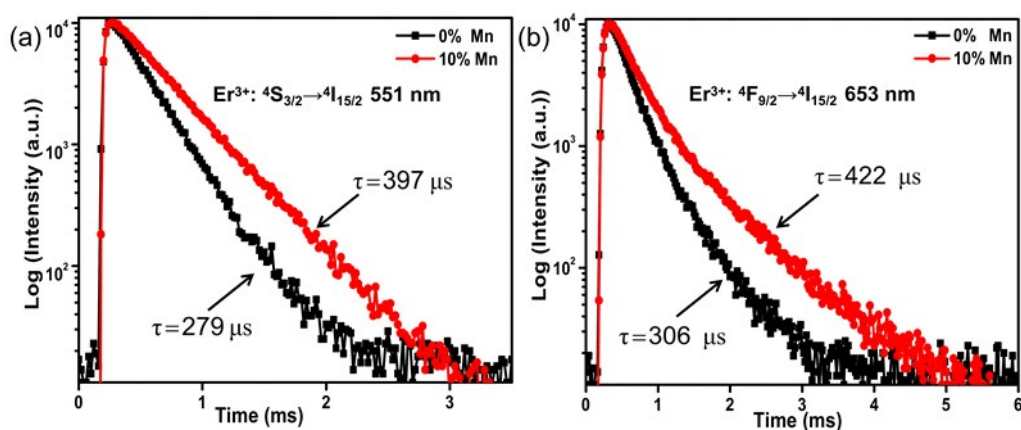


Figure S5. Upconversion luminescence decay curves of Er^{3+} emission at 551 and 653 nm from the $\text{LiYF}_4:\text{Yb}/\text{Er}$ nanoparticles with 0 and 10 mol% Mn doping, under pulsed laser excitation at 980 nm.

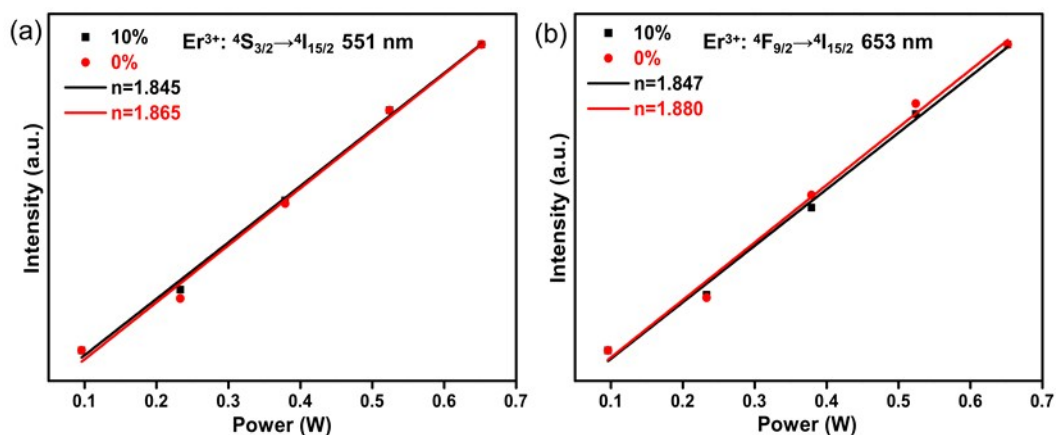


Figure S6. The excitation power dependence of green and red UC emission of $\text{LiYF}_4:\text{Yb}/\text{Er}$ (20/2 mol%) NPs with 0 and 10 mol% Mn doping (n represents the numerical value of the slope determined from the linear fitting of the experimental results).

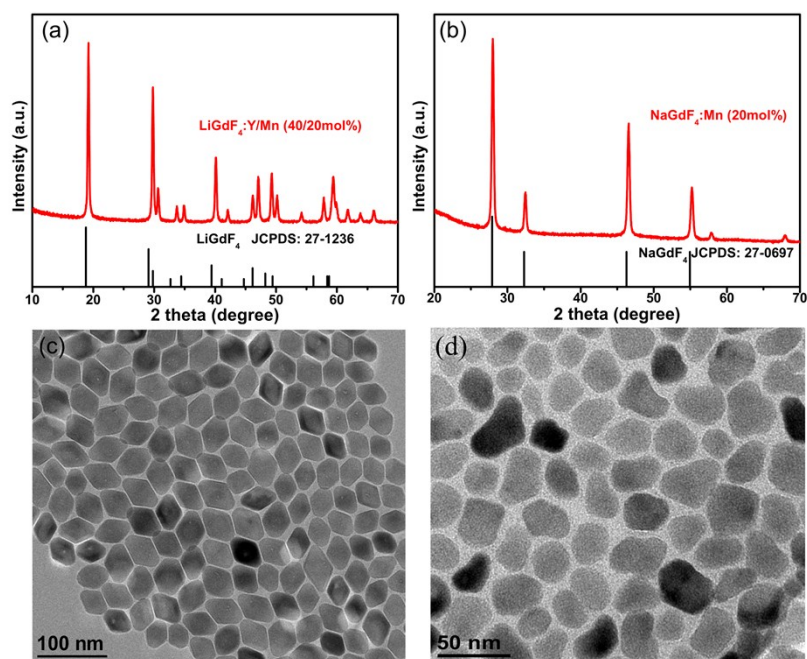


Figure S7. (a-b) XRD patterns and (c-d) TEM images of $\text{LiYF}_4\text{:Gd/Mn (40/20 mol\%)}$ and $\text{NaGdF}_4\text{:Mn (20 mol\%)}$ nanoparticles.

Table S3. Elemental analyses of Mn^{2+} in $\text{LiYF}_4\text{:Gd/Mn}$ (40/20mol%) and $\text{NaGdF}_4\text{:Mn}$ (10-30 mol%) by ICP-AES measurement.

Sample	Nominal concentration (mol%)	ICP-AES result (mol%)
$\text{LiYF}_4\text{:Gd/Mn}$ (40/20mol%)	20	1.78
$\text{NaGdF}_4\text{:Mn}$ (30 mol%)	30	2.02
$\text{NaGdF}_4\text{:Mn}$ (20 mol%)	20	1.83
$\text{NaGdF}_4\text{:Mn}$ (10 mol%)	10	0.64

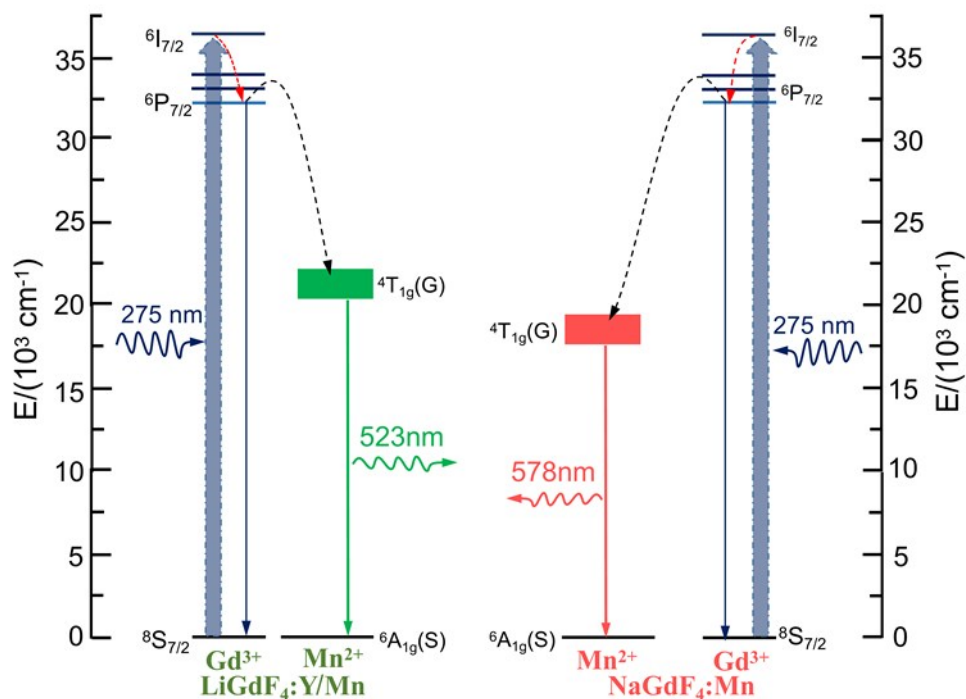


Figure S8. The dominant energy transfer processes in the $\text{LiYF}_4:\text{Gd/Mn}$ (40/20 mol%) (left) and $\text{NaGdF}_4:\text{Mn}$ (20 mol%) (right) nanoparticles.

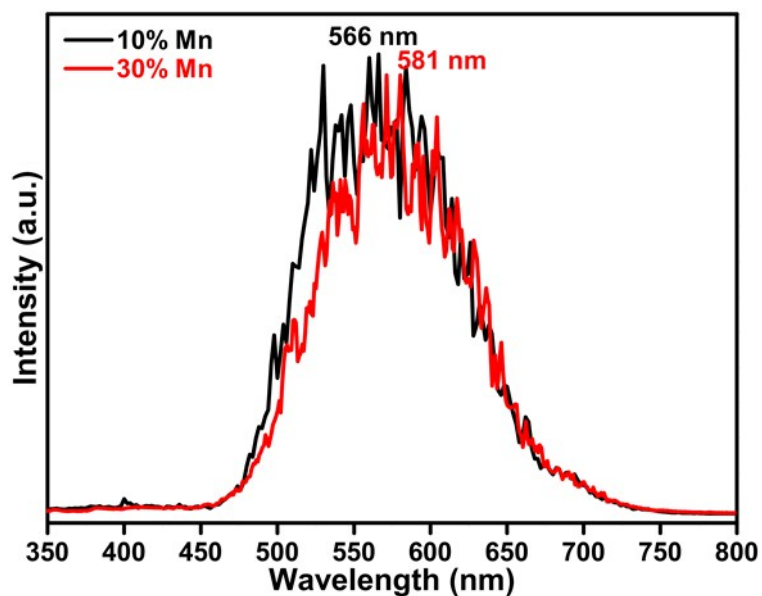


Figure S9. The downconversion emission spectra of the $\text{NaGdF}_4:\text{Mn}$ (10 and 30 mol%) colloidal solutions under excitation at 275 nm.

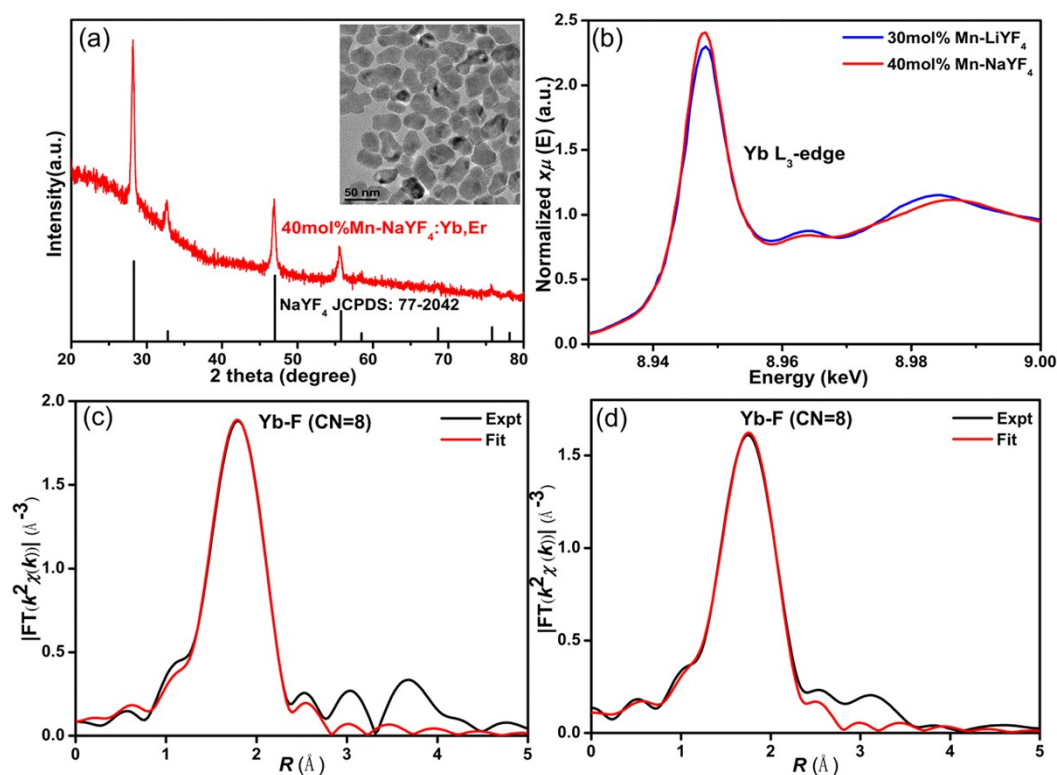


Figure S10. (a) XRD pattern of the as-synthesized NaYF₄:Yb/Er nanoparticles doped with 40 mol% Mn²⁺ ions and the standard diffraction pattern of cubic-phase NaYF₄ (JCPDS No. 077-2042). Inset shows the TEM image of the corresponding nanoparticles. (b) Yb L₃-edge XANES spectra recorded for 30 mol% Mn-LiYF₄:Yb/Er and 40 mol% Mn-NaYF₄:Yb/Er nanoparticles. FT k³-weighted $\chi(k)$ -function of EXAFS spectrum for Yb³⁺ in 30 mol% Mn-LiYF₄:Yb/Er (c) and 40 mol% Mn-NaYF₄:Yb/Er (d) with Yb-F fitting.

Table S4. Yb L₃-edge and Mn K-edge EXAFS curve fitting parameters^a

sample	chemical bond	CN	R (Å)	σ^2 (Å ²)	ΔE_0 (eV)
30mol%Mn-LiYF ₄ ^b	Yb-F	8.9	2.24	0.007	2.2
	Mn-F	4.1	2.06	0.004	-6.5
40mol%Mn-NaYF ₄ ^b	Yb-F	8.6	2.22	0.010	1.7
	Mn-F	6.1	2.06	0.004	-5.4

^aCN is the coordination number; R is interatomic distance; σ^2 is Debye-Waller factor (a measure of thermal and static disorder in absorber-scatterer distances); ΔE_0 is the edge-energy shift. Error bounds of the structural parameters were estimated as $N \pm 20\%$; $R \pm 1\%$; $\sigma^2 \pm 20\%$; $\Delta E_0 \pm 20\%$. S_0^2 were fixed to 0.77 for Yb L₃-edge fitting and 0.7 for Mn K-edge fitting. ^bYb L₃-edge fitting range: $2.0 \leq k$ (1/Å) ≤ 8.5 and $1.0 \leq R$ (Å) ≤ 2.4 ; Mn K-edge fitting range: $2.0 \leq k$ (1/Å) ≤ 7.0 and $0.8 \leq R$ (Å) ≤ 2.2 .

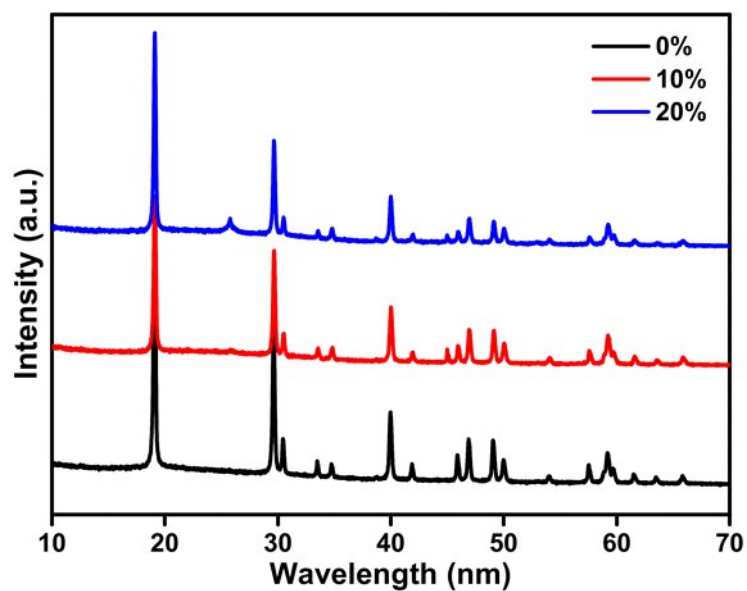


Figure S11. XRD patterns of the as-synthesized LiYF₄:Yb/Tm (25/0.5 mol%) nanoparticles obtained in the presence of 0, 10 and 20 mol% Mn²⁺ dopant ions, respectively.

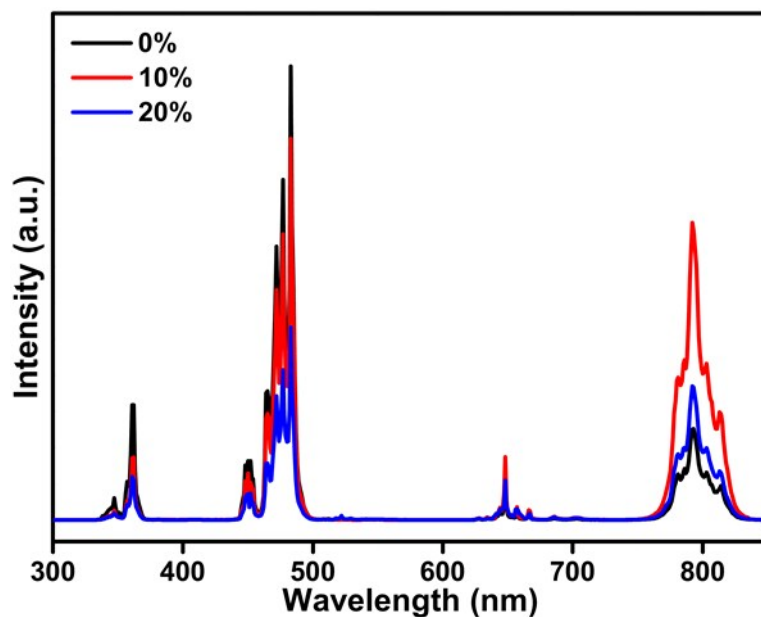


Figure S12. Room-temperature UC emission spectra of cyclohexane solutions containing LiYF₄:Yb/Tm/Mn (25/0.5/0-20 mol%) nanoparticles with different Mn²⁺ doping levels, under 980 nm laser excitation (CW, 2 W·cm⁻²).

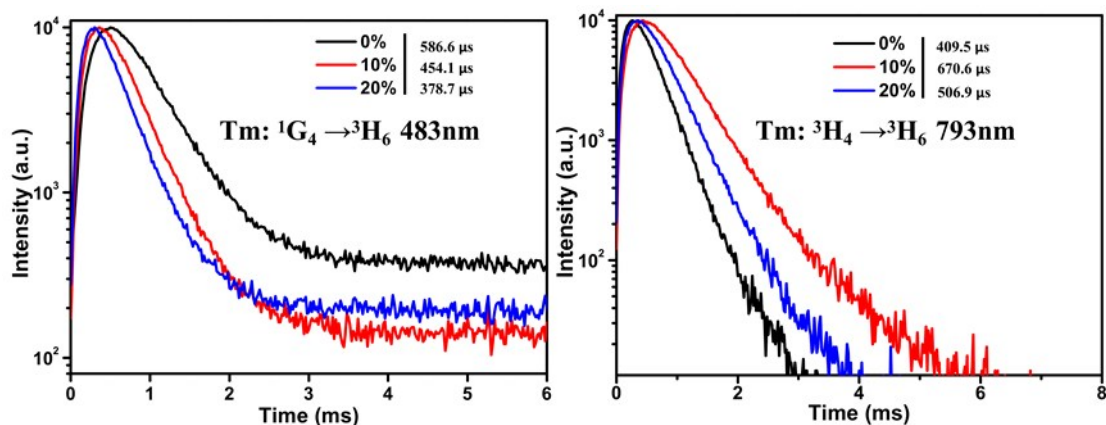


Figure S13. Upconversion luminescence decay curves of Tm³⁺ emission at 483 and 793 nm from the LiYF₄:Yb/Tm nanoparticles with 0, 10 and 20 mol% Mn doping, under pulsed laser excitation at 980 nm.

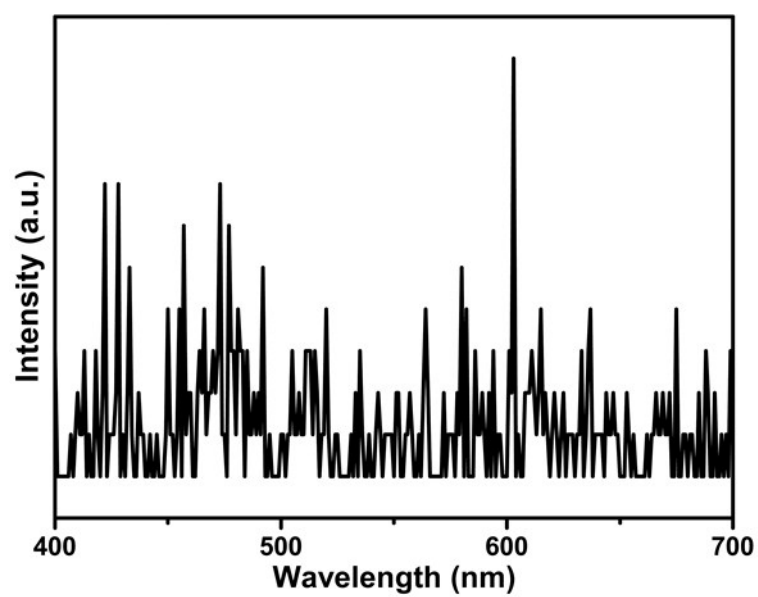


Figure S14. The emission spectra of cyclohexane solutions containing LiYbF₄:Mn (10 mol%) nanoparticles at 4K under 980 nm laser excitation.

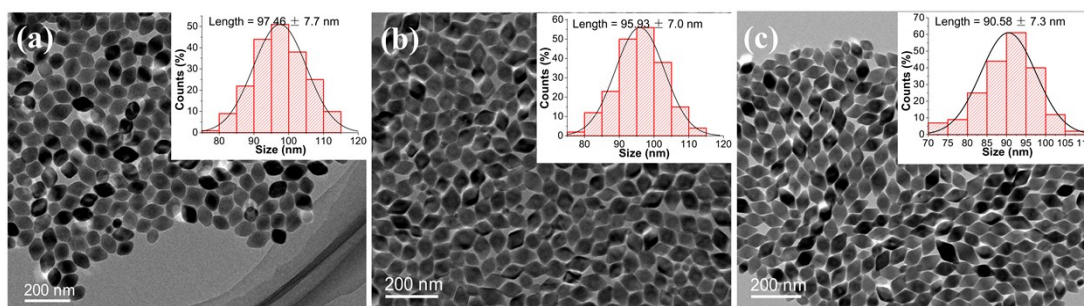


Figure S15. Low-resolution TEM images and size distributions of (a) LiYF₄:Er (10mol%), (b) LiYF₄:Er/Mn (10/10 mol%) and (c) LiYF₄:Er/Mn (10/20 mol%) nanocrystals. The size distributions of the nanocrystals were calculated by counting over 200 particles recorded in the TEM images.

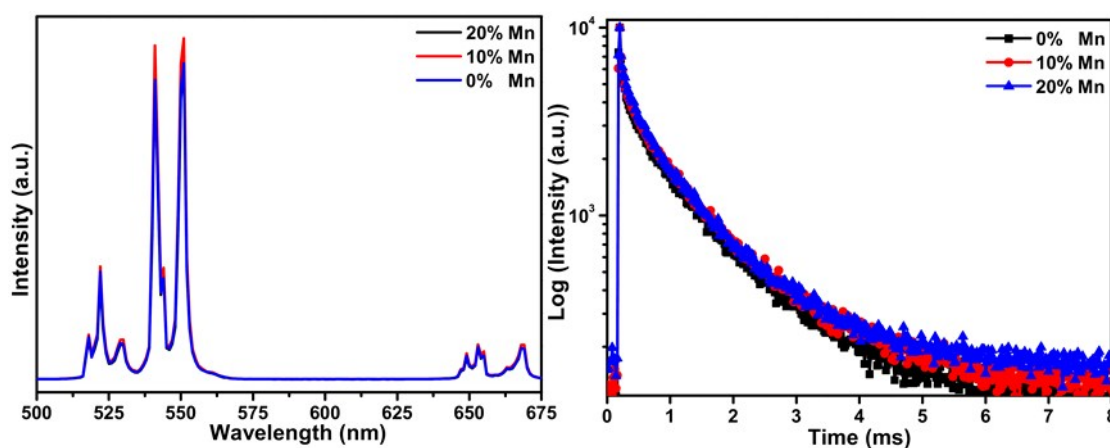


Figure S16. (a) Emission spectra of the LiYF₄:Er/Mn (10/0-20 mol%) nanocrystals, under the excitation of 980nm CW laser (2W·cm⁻²). (b) Upconversion luminescence decay curves of Er³⁺ emissions at 551 nm for the LiYF₄:Er/Mn (10/x mol%) nanocrystals. Remarkably, the emission intensities and lifetimes of Er³⁺ were found to be independent of Mn²⁺ concentration.

Notes and references

- 1 D. M. Roy, R. Roy, *J. Electrochem. Soc.*, 1964, **111**, 421-429.
- 2 A. V. Goryunov, A. I. Popov, N. M. Khajdukov and P.P.Fedorov, *Mater. Res. Bull.*, 1992, **27**, 213-220.