

SUPPORTING INFORMATION

for

The Forest or the Trees: A Critical Review on the Analysis of Total Organic Halogen (TOX) in Drinking Waters and Its Utility as Water Quality Parameter

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Table S1 Reported recoveries of TOX analysis (adsorption-combustion-detection) of model halogenated compounds

Analyte	% Recovery	Log K _{ow}	Adsorbent	Reference
<i>Trihalomethanes</i>				
Chloroform	90.5	1.97 ^a	Sugar activated carbon (Merck)	1
	94		Activated carbon	2
	89		Filtrisorb 400 and APC carbon (Calgon)	3
	93 ^b ; 84 ^c		Activated carbon	4
	95 ± 11		Activated carbon (CPI International)	5
Bromodichloromethane	98	1.61 ^f	Activated carbon	2
	102 ± 1 ^d ; 101 ± 4 ^e		Activated carbon (CPI International)	6
	TOCl: 118 ± 0.6; TOBr: 120 ± 2		Activated carbon (CPI International)	5
Chlorodibromomethane	86	2.16 ^a	Activated carbon	2
	103 ± 3 ^d ; 104 ± 2 ^e		Activated carbon (CPI International)	6
	TOCl: 120 ± 17; TOBr: 92 ± 2		Activated carbon (CPI International)	5
Bromoform	95.3 ^b ; 97.7 ^c	2.40 ^f	Activated carbon	4
	90.6		Sugar activated carbon (Merck)	1
	98 ± 5a		Activated carbon (CPI International)	6
	98 ± 4b		Activated carbon (CPI International)	6
	101		Filtrisorb 400 and APC carbon (Calgon)	3
	101		Activated carbon	2
	119 ± 2		Activated carbon (CPI International)	5
Iodoform	100 ± 2	3.03 ^g	Activated carbon (CPI International)	5
<i>Haloacetic acids</i>				
Chloroacetic acid	56.0	0.2 ^h	Sugar activated carbon (Merck)	1
	91 ± 3 ^d ; 92 ± 4 ^e		Activated carbon (CPI International)	6
	0		Filtrisorb 400 and APC carbon (Calgon)	3
	35		Activated carbon (Mitsubishi)	7

	96 ± 2 78 ± 6		Activated carbon (CPI International)	5
			Activated carbon (CPI International)	8
Bromoacetic acid	96 ± 2 ^d ; 95 ± 1 ^e 69 95 ± 4	0.41 ^a	Activated carbon (CPI International)	6
			Activated carbon (Mitsubishi)	7
			Activated carbon (CPI International)	8
Bromochloroacetic acid	97 ± 5 ^d ; 100 ± 5 ^e TOCl: 109 ± 2; TOBr: 110 ± 0.6 TOCl: 94 ± 4; TOBr: 111 ± 4	0.61 ^f	Activated carbon (CPI International)	6
			Activated carbon (CPI International)	5
				8
Dichloroacetic acid	98 ± 4 ^d ; 101 ± 5 ^e 110 ± 12	0.942 ^h	Activated carbon (CPI International)	6
			Activated carbon (CPI International)	8
Dibromoacetic acid	102 ± 3 ^d ; 101 ± 4 ^e 95 ± 6	0.70 ^f	Activated carbon (CPI International)	6
			Activated carbon (CPI International)	8
Chlorodibromoacetic acid	TOCl: 80 ± 10; TOBr: 97 ± 2	1.62 ^a	Activated carbon (CPI International)	8
Bromodichloroacetic acid	TOCl: 87 ± 6; TOBr: 92 ± 8	1.53 ^a	Activated carbon (CPI International)	8
Trichloroacetic acid	109 83 100 ± 2 ^d ; 97 ± 4 ^e 85 ± 4	1.645 ^h	Sugar activated carbon (Merck)	1
			Activated carbon (Mitsubishi)	7
			Activated carbon (CPI International)	6
			Activated carbon (CPI International)	8
Tribromoacetic acid	98 ± 3 ^d ; 98 ± 5 ^e 108 ± 0.9	1.71 ^f	Activated carbon (CPI International)	6
			Activated carbon (CPI International)	5
Iodoacetic acid	103 99 ± 3 ^d ; 94 ± 5 ^e	0.85 ^f	Activated carbon (Mitsubishi)	7
			Activated carbon (CPI International)	6
<i>Haloacetonitriles</i>				
Chloroacetonitrile	85 ± 10 80 ± 8	0.11 ^g	Activated carbon (CPI International)	5
			Activated carbon (CPI International)	8
Bromoacetonitrile	127 ± 2 108 ± 2	0.20 ^g	Activated carbon (CPI International)	5
			Activated carbon (CPI International)	8

Dichloroacetonitrile	105 ± 4	0.29 ^h	Activated carbon (CPI International)	8
Dibromoacetonitrile	107 ± 6	0.47 ^h	Activated carbon (CPI International)	8
Bromochloroacetonitrile	TOCl: 135 ± 15; TOBr: 118 ± 10	0.38 ^h	Activated carbon (CPI International)	8
Trichloroacetonitrile	124 ± 2	1.21 ^g	Activated carbon (CPI International)	5
	115 ± 10		Activated carbon (CPI International)	8
Halophenols				
2-Chlorophenol	101 ± 6	2.16 ^h	Activated carbon (CPI International)	8
4-Chlorophenol	90.7	2.39 ^h	Sugar activated carbon (Merck)	1
	99		Activated carbon (Mitsubishi)	7
	89 ± 4		Activated carbon (CPI International)	8
2,4-Dichlorophenol	98	3.17 ^a	Activated carbon (Calgon F400)	9
2,6-Dichlorophenol	93 ± 8	2.8 ^h	Activated carbon (CPI International)	8
2,4,6-Trichlorophenol	96.8 ^b ; 96.8 ^c	3.69 ^a	Activated carbon	4
	101		Filtrisorb 400 and APC carbon (Calgon)	3
	104 ± 4		Activated carbon (CPI International)	5
	91 ± 5		Activated carbon (CPI International)	8
Pentachlorophenol	93	5.86 ^g	Activated carbon	2
	82 ± 12		Activated carbon (CPI International)	8
4-Bromophenol	97	2.33 ^a	Activated carbon (Calgon F400)	9
	99.2		Sugar activated carbon (Merck)	1
	96		Activated carbon (Mitsubishi)	7
	99 ± 2		Activated carbon (CPI International)	8
2,4-Dibromophenol	108 ± 0.9	3.29 ^h	Activated carbon (CPI International)	5
	83 ± 6		Activated carbon (CPI International)	8
2,4,6-Tribromophenol	90.6	4.08 ^a	Activated carbon	4
	102		Filtrisorb 400 and APC carbon (Calgon)	3
	102 ± 4		Activated carbon (CPI International)	5

	91 ± 2		Activated carbon (CPI International)	8
2-Iodophenol	85 ± 1	2.68 ^h	Activated carbon (CPI International)	8
4-Iodophenol	98	2.90 ^a	Activated carbon (Mitsubishi)	7
	96 ± 4		Activated carbon (CPI International)	5
Halobenzenes				
4-Amino-chlorobenzene	80	1.83 ^a	Activated carbon (Mitsubishi)	7
4-Amino-bromobenzene	88	2.26 ^h	Activated carbon (Mitsubishi)	7
4-Amino-iodobenzene	80	2.80 ^a	Activated carbon (Mitsubishi)	7
Chlorobenzene	67.9	2.89 ^h	Sugar activated carbon (Merck)	1
1,2-Dichlorobenzene	40.6	3.38 ^a	Sugar activated carbon (Merck)	1
	94 ± 4		Activated carbon (CPI International)	8
1,3-Dichlorobenzene	107	3.53 ^f	Filtrisorb 400 and APC carbon (Calgon)	3
1,4-Dichlorobenzene	55	3.45 ^a	Activated carbon (Mitsubishi)	7
1,3,5-Trichlorobenzene	83.9	4.15 ^a	Sugar activated carbon (Merck)	1
Bromobenzene	79.8	2.99 ^h	Sugar activated carbon (Merck)	1
	95		Filtrisorb 400 and APC carbon (Calgon)	3
	113 ± 6		Activated carbon (CPI International)	5
	109 ± 6		Activated carbon (CPI International)	8
1,2-Dibromobenzene	94 ± 4	3.28 ^a	Activated carbon (CPI International)	8
1,4-Dibromobenzene	46.1	3.79 ^h	Sugar activated carbon (Merck)	1
	49		Activated carbon (Mitsubishi)	7
	93 ± 3		Activated carbon (CPI International)	5
	98 ± 2		Activated carbon (CPI International)	8
2-Bromoethylbenzene	56.4	3.783 ^h	Sugar activated carbon (Merck)	1
1,4-Diiodobenzene	78	4.11 ^a	Activated carbon (Mitsubishi)	7
4-Bromobenzylbromide	28.4	3.77 ^f	Sugar activated carbon (Merck)	1
1-Bromonaphthalene	57.1	4.35 ^h	Sugar activated carbon (Merck)	1

1-Bromonitrobenzene	58.4	2.5 ^h	Sugar activated carbon (Merck)	1
4-Bromotoluene	93.4	3.42 ^f	Sugar activated carbon (Merck)	1
Haloamides				
2-Chloroacetamide	14	-0.53 ^a	Activated carbon (Mitsubishi)	7
2,2,2-Trichloroacetamide	121 ± 2	0.83 ^g	Activated carbon (CPI International)	5
3-Chloropropionamide	79.3	-0.42 ^f	Sugar activated carbon (Merck)	1
	107 ± 2		Activated carbon (CPI International)	5
2-Iodoacetamide	37	-0.19 ^a	Activated carbon (Mitsubishi)	7
X-ray contrast media (iodinated organic compounds)				
Iopromide ^h	98	-2.05 ^a	Activated carbon (Mitsubishi)	7
	98 ± 2		Activated carbon (CPI International)	5
Iopamidol	99 + 0.1		Activated carbon (CPI International)	5
Iothalamic acid	112 ± 4	-0.056 ^g	Activated carbon (CPI International)	5
Iodipamide	97 ± 5	5.10 ^g	Activated carbon (CPI International)	5
Iohexol	100 ± 0.2		Activated carbon (CPI International)	5
Diatrizoic acid	94	0.49 ^g	Activated carbon (Mitsubishi)	7
	104 ± 4		Activated carbon (CPI International)	5
Haloketones				
Chloroacetone	90	0.02 ^f	Filtrisorb 400 and APC carbon (Calgon)	3
Halobenzoic acids				
3-Chlorobenzoic acid	100 ± 2	2.52 ^g	Activated carbon (CPI International)	5
4-Chlorobenzoic acid	83.7	2.65 ^h	Sugar activated carbon (Merck)	1
3-Bromobenzoic acid	104	2.75 ^a	Filtrisorb 400 and APC carbon (Calgon)	3
4-Bromobenzoic acid	96.2	2.86 ^h	Sugar activated carbon (Merck)	1
3,5-Dichloroisonicotinic acid	100 ± 2 ^d ; 101 ± 3 ^e	1.62 ⁱ	Activated carbon (CPI International)	6

<i>Haloalkanes</i>				
Dichloromethane	98.1 50 ± 9	1.25 ^h	Sugar activated carbon (Merck) Activated carbon (CPI International)	1 5
1-Chloro-2-Bromoethane	106	1.92 ^f	Filtrisorb 400 and APC carbon (Calgon)	3
1,1,1-Trichloroethane	92.1	2.49 ^h	Sugar activated carbon (Merck)	1
1,1-Dibromoethane	64.4	1.94 ^f	Sugar activated carbon (Merck)	1
1-Bromo-3-chloropropane	87.0	2.2 ^h	Sugar activated carbon (Merck)	1
1-Chlorohexane	82.5	3.58 ^h	Sugar activated carbon (Merck)	1
<i>Haloalkenes</i>				
Tetrachloroethylene	47.9	3.40 ^h	Sugar activated carbon (Merck)	1
γ-Hexachlorocyclohexane	105.0	4.14 ^f	Sugar activated carbon (Merck)	1
4-Bromo-1-butene	69.1	2.34 ^a	Sugar activated carbon (Merck)	1
<i>Other halogenated nitrogen-containing compounds</i>				
Chloramine-T (Cl ₂)	3.4	1.85 ⁱ	Sugar activated carbon (Merck)	1
Tribromonitromethane	114 ± 8	1.59 ^g	Activated carbon (CPI International)	5
4-Chloroaniline	102.6 110 ± 3	1.83 ^a	Sugar activated carbon (Merck) Activated carbon (CPI International)	1 5
2-Bromoaniline	95.4	2.11 ^h	Sugar activated carbon (Merck)	1
<i>Halogenated sulphur containing compounds</i>				
2-Bromothiophene	42.1	2.75 ^a	Sugar activated carbon (Merck)	1
3',3''-Dibromophenol-sulfonphthaleine	98		Activated carbon (Mitsubishi)	7
3', 3''-Dichlorophenol-sulfonphthaleine	92		Activated carbon (Mitsubishi)	7
<i>Other model compounds</i>				
Bis(2-chloroethyl)ether	92	1.29 ^a	Activated carbon	2
2-Chloroethanol	20	0.107 ^j	Filtrisorb 400 and APC carbon (Calgon)	3

	6.6 50 ± 2		Sugar activated carbon (Merck) Activated carbon (CPI International)	1 5
(6-Bromo-pyridin-2-yl) methanol	101 ± 2 ^d ; 99 ± 4 ^e	0.994 ^j	Activated carbon (CPI International)	6

^a <http://logKow.cisti.nrc.ca/logKow/search.html>

^b analysis done by Mitsubishi TOX Analyser

^c analysis done by Dohrmann TOX Analyser

^d detection by microcoulometer

^e detection by ion chromatograph

^f <http://www.chemspider.com>

^g http://www.epa.gov/oswer/riskassessment/ragse/pdf/appendix_a.pdf

^h <https://pubchem.ncbi.nlm.nih.gov/compound>

ⁱ <http://www.chemicalize.org/structure>

^j SciFinder Website: Under 'Explore the substances'

References

1. G. Brandt and A. Kettrup, Determination of organic group parameters (AOCl, AOBr, AOS) in water by means of ion-chromatographic detection. *Fresenius J. Anal. Chem.*, 1987, **327**, 213-219.
2. R. Dressman, B. Najar and R. Redzikowski, Proceedings of the AWWA Water Quality and Technology Conference, Philadelphia, 1979, The analysis of organohalides (OX) in water as a group parameter.
3. Y. Takahashi, R. T. Moore and R. J., Joyce, in *Chemistry in Water Reuse Vol. 2*, ed. W. J. Cooper, Ann Arbor Science, Michigan, 1981. Measurement of total organic halides (TOX) and purgeable organic halides (POX) in water using carbon adsorption and microcoulometric determination.
4. D.A. Reckhow, C. Hull, E. Lehan, J.M. Symons, H-S. Kim, Y-M.Chang, L. Simms, R.C. Dressman and H. Pourmoghaddas, The determination of total organic halide in water: A comparative study of two instruments, *Int. J. Environ. Anal. Chem.*, 1990, **38**, 1-7.

5. I. Kristiana, S. McDonald, J. Tan, C. A. Joll and A. Heitz, Analysis of halogen-specific TOX revisited: Method improvement and application, *Talanta*, 2015, **139**, 104-110.
6. G. Hua and D. A. Reckhow, Determination of TOCl, TOBr, and TOI in drinking water by pyrolysis and off-line ion chromatography. *Anal. Bioanal. Chem.*, 2006, **384**, 495-504.
7. J. Oleksy-Frenzel, S. Wischnack and M. R. Jekel, Application of ion chromatography for the determination of the organic group parameters AOCl, AOBr, and AOI in water. *Fresenius J. Anal. Chem.*, 2000, **366**, 89-94.
8. M. Langsa, S. Allard, I. Kristiana, A. Heitz and C. A. Joll, Halogen-specific total organic halogen analysis: Assessment by recovery of total bromine. *J. Environ. Sci.*, 2017, **58**, 340-348.
9. M. R. Jekel and P. V. Roberts, Total organic halogen as a parameter for the characterization of reclaimed waters: Measurement, occurrence, formation, and removal, *Environ. Sci. Technol.*, 1980, **14**, 970 – 975.

Table S2: Summary of outcomes from studies reporting toxicity of TOX in disinfected water

Sample/Disinfection conditions	[TOX]	Bioassay method	Cell/organism type	Toxicological endpoint	Toxicity measure	Ref
Water samples from various locations in a water treatment plant (WTP)	31.8 – 325 $\mu\text{g L}^{-1}$	Microtox	<i>Vibrio fischeri</i>	Cytotoxicity	EC ₅₀ (REF): 2.04 – 17.25 TEQ (mg L ⁻¹): 0.71 – 6.61	1
		Mammalian cell cytotoxicity	Human colon cancer cell line (Caco-2)	Cytotoxicity	Assay was insensitive	
		UmuC assay (without metabolic activation) UmuC assay (with metabolic activation)	<i>Salmonella typhimurium</i>	Genotoxicity	EC _{IR1.5} – S9 (REF): 16 – 198 4NQO-EQ – S9 ($\mu\text{g L}^{-1}$): 0.05 – 0.72	
			<i>Salmonella typhimurium</i>		EC _{IR1.5} + S9 (REF): 26 – 244 2AA-EQ ($\mu\text{g L}^{-1}$) + S9: 0.18 – 1.63	
			<i>E.coli</i> GSH ₊	Reactive toxicity to protein	Ratio EC ₅₀ (GSH ₊) : EC ₅₀ (GSH ₋): 3.09 – 3.68	
			<i>E.coli</i> biosensor			
Drinking water samples from various WTPs and tap waters; various sample enrichment methods for bioassays	1.3 – 2.8 μM	AREc32 assay	AREc32 cell	Oxidative stress	EC _{IR1.5} (REF): 2 – 35 tBHQ-EQ (mg L ⁻¹): 0.01 – 0.11	2
		Microtox	<i>Aliivibrio fischeri</i>	Cytotoxicity	EC ₅₀ (REF): 13.9 – 26.0	
Disinfection of treated water samples from 3 WTPs Chlorination or chloramination, pH 7, 72hr contact time	0.3 – 8 μM	AREc32 assay	AREc32 cell	Oxidative stress	EC _{IR1.5} (REF): 1.5 – 2.2	3
		Microtox	<i>Vibrio fischeri</i>	Bacterial cytotoxicity	EC ₅₀ (REF): 2.40 – 51.3 TEQ (mg L ⁻¹): 0.24 – 5.22	
		UmuC assay (without metabolic activation)	<i>Salmonella typhimurium</i>	Genotoxicity	EC _{IR1.5} – S9: 14 – >150 4NQO-EQ – S9 ($\mu\text{g L}^{-1}$): <0.06 – 0.87	
		UmuC assay (with metabolic activation)	<i>Salmonella typhimurium</i>	Genotoxicity	EC _{IR1.5} + S9: 27 – >150 2AA-EQ + S9 ($\mu\text{g L}^{-1}$): <0.71 – 4.05	
		<i>E.coli</i> biosensor	<i>E.coli</i> GSH ₊	Reactive toxicity to protein	Ratio of EC ₅₀ (GSH ₊) : EC ₅₀ (GSH ₋): 0.7 – 6.9	
		AREc32 assay	AREc32 cell	Oxidative stress	EC _{IR1.5} : 1.8 – 50 tBHQ-EQ (mg L ⁻¹): <5 – 123	

Disinfection of NOM solution using MP UV-Cl ₂ /NH ₂ Cl, pH 7, 24hr contact time	95 – 859 µM as Cl	<i>In vitro</i> chronic cytotoxicity assay	Normal human colon cells	Cytotoxicity	IC ₅₀ : 80 – 564 µM as Cl (inhibitory concentration at which 50% reduction in cell density occurred)	4
Chlorination and chloramination of treated water sample from a WTP (with and without addition of bromide/iodide), pH 8, 48hr contact time. Organic materials extracted by XAD resin	TOCl: 500 – 3000 µg L ⁻¹ TOBr: 500 – 6000 µg L ⁻¹ TOI: 8 – 500 µg L ⁻¹	Cytotoxicity assay	CHO cell	Cytotoxicity	Cytotoxicity index (CTI): 12 – 40	5
		Single cell gel electrophoresis (SCGE) assay	CHO cell	Genotoxicity	Genotoxicity index (GTI): 10 – 65	
Chlorination of treated water samples, 4 hr contact time	52 – 881 µg L ⁻¹	Cytotoxicity assessment (cell counting kit-8)	Human hepatocyte cell line (HepG2)	Cytotoxicity	Cytotoxicity index: 1.3 – 2.7	6
		Single cell gel electrophoresis (SCGE) assessment	HepG2 cells	Genotoxicity	Olive Tail Moment: 1.7 – 40	
		Oxidative stress assessment	HepG2 cells	Oxidative stress	Total superoxide dismutase (T-SOD): 0.7 – 2.3 nmol/mg protein Glutathione (GSH): 220 – 360 nmol/mg protein Malondialdehyde (MDA): 0.3 – 3.5 nmol/mg protein	
Chlorinated simulated tap water samples, pH 8.5, 15hr contact time	0.2 – 0.7 mg L ⁻¹ as Cl	Chronic cytotoxicity assay	CHO cells	Cytotoxicity	Cytotoxicity index: 3.1 – 13.5	7
Chlorination and bromination of humic acid solution, pH 7, 24 hr contact time	14.4 – 185.7 µg Cl L ⁻¹	Chromosomal aberration test	Chinese hamster lung (CHL) cells	Mutagenicity	No. of abnormal chromosomes per TOX: 9.9 – 56.7 count/100 cells mmol	8
NOM solution treated with UV/Cl ₂	10 – 17 µM	Cytotoxicity assay	Chinese hamster ovary (CHO) cells	Cytotoxicity	20 – 35 mg-phenol L ⁻¹	9

Disinfection of simulated raw water (Cl ₂ , NH ₂ Cl, ClO ₂ , O ₃ +Cl ₂) 48 hr contact time	5 – 550 µg L ⁻¹ as Cl	Developmental toxicity bioassay	<i>Platynereis dumerilii</i> embryos	Developmental toxicity	2 – 95% normal development (presented in graphs)	10
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References

1. P. A. Neale, A. Antony, M. E. Bartkow, M. J. Farré, A. Heitz, I., Kristiana, J. Y. M. Tang and B. I. Escher, Bioanalytical assessment of the formation of disinfection byproducts in a drinking water treatment plant. *Environ. Sci. Technol.*, 2012, **46**, 10317-10325.
2. D. Stalter, L. I. Peters, E. O'Malley, J. Y. M. Tang, M. Revalor, M. J. Farré, K. Watson, U. von Gunten and B. Escher, Sample enrichment for bioanalytical assessment of disinfected drinking water: Concentrating the polar, the volatiles, and the unknowns, *Environ. Sci. Technol.*, 2016, **50**, 6495-6505.
3. M. J. Farré, S. Day, P. A. Neale, D. Stalter, J. Y. M. Tang and B. I. Escher, Bioanalytical and chemical assessment of the disinfection by-product formation potential: Role of organic matter, *Water Res.*, 2013, **47**, 5409-5421.
4. B. A. Lyon, R. Y. Milsik, A. B. DeAngelo, J. E. Simmons, M. P. Moyer and H. S. Weinberg, Integrated chemical and toxicological investigation of UV-Chlorine/Chloramine drinking water treatment. *Environ. Sci. Technol.*, 2014, **48**, 6743-6753.
5. Y., Yang, Y., Komaki, S. Y., Kimura, H-Y., Hu, E. D., Wagner, B. J., Marinas and M. J. Plewa, Toxic impact of bromide and iodide on drinking water disinfected with chlorine or chloramines, *Environ. Sci. Technol.*, 2014, **48**, 12362-12369.
6. P. Shi, R. Ma, Q. Zhou, A. Li, B. Wu, Y. Miao, X. Chen and X. Zhang, Chemical and bioanalytical assessments on drinking water treatments by quaternized magnetic microspheres, *J. Hazard. Mater.*, 2015, **285**, 53-60.
7. Y. Pan, X. Zhang, E. D. Wagner, J. Osiol and M. J. Plewa, Boiling of simulated tap water: Effect on polar brominated disinfection by-products, halogen speciation, and cytotoxicity, *Environ. Sci. Technol.*, 2014, **48**, 149-156.
8. S. Echigo, S. Itoh, T. Natsui, T. Araki and R. Ando, Contribution of brominated organic disinfection by-products to the mutagenicity of drinking water, *Water Sci. Technol.*, 2004, **50**(5), 321-328.
9. W-L. Wang, X. Zhang, Q-Y. Wu, Y. Du, and H-Y. Hu, Degradation of natural organic matter by UV/chlorine oxidation: Molecular decomposition, formation of oxidation byproducts and cytotoxicity, *Water Res.*, 2017, **124**, 251-258.

10. J. Han and X. Zhang, Evaluating the comparative toxicity of DBP mixtures from different disinfection scenarios: A new approach by combining freeze-drying or rotoevaporation with a marine polychaete bioassay, *Environ. Sci. Technol.*, 2018, **52**, 10552-10561.